

RESEARCH ARTICLE

4-(4-Morpholinophenyl)-6-arylpyrimidin-2-amines: synthesis, spectral analysis, and in vitro microbiological evaluation

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Abstract

Compounds 4-(4-morpholinophenyl)-6-phenylpyrimidin-2-amine **20**, 4-(4-methoxyphenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine **23**, 4-(4-bromophenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine **25**, 4-(3-chlorophenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine **27**, and 4-(3-fluorophenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine **28** exerted excellent antibacterial activity against *V. cholerae*. Compounds 4-(4-chlorophenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine **22**, **25**, 4-(4-morpholinophenyl)-6-(3-nitrophenyl)pyrimidin-2-amine **26**, and **28**, which contained electron-withdrawing chloro, bromo, nitro, and fluoro functional groups, respectively, at the *para/meta* position of the phenyl ring attached to the pyrimidine ring promoted much activity against *S. aureus*. Compounds 4-(4-fluorophenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine **24** and **25** (against β -hemolytic *Streptococcus*) and compound **28** (against *S. felxneri*) showed pronounced activity. Compounds **26** and **28** (against *K. pneumoniae*) and compounds **24**, **25**, and **28** (against *P. aeruginosa*) exerted strong antibacterial activity. Compounds **22** and **25** promoted much antifungal activity against *A. flavus*, while compounds **24** and **25** registered maximum activity against *Mucor*. Compounds **23** (against *C. albicans*) and **27** and **28** (against *Rhizopus*) promoted good activity.

Keywords: 4-(4-morpholinophenyl)-6-arylpyrimidin-2-amines; guanidine nitrate; synthesis; antibacterial activity; antifungal activity

Introduction

Pyrimidines constitute the basic nucleus in nucleic acids and have been associated with a number of biological activities. Some notable biological activities of pyrimidine derivatives include as adenosine receptor antagonists¹, as kinase inhibitors², analgesic³, anti-inflammatory³, as inhibitors of cyclin-dependent kinases 1 and 2⁴, as calcium channel antagonists⁵, antihistaminic⁶, and antitubercular⁷ activities. Substituted aminopyrimidine nuclei are common in marketed drugs such as anti-atherosclerotic aronixil, anti-histaminic thonzylamine, anti-anxiolytic buspirone, anti-psoriatic enazadrem, and other medicinally relevant compounds.

Morpholine is a simple heterocyclic compound with great industrial importance. It is used as an anticorrosive agent and as a chemical intermediate: catalyst, solvent,

and antioxidant, in the production of various pharmaceuticals and pesticides. 4-Phenyl morpholine derivatives⁸ are reported to possess antimicrobial, anti-inflammatory, and central nervous system activities. Linezolid (commercially available antimicrobial) also possesses a 4-phenyl-morpholine substituent. These derivatives are reported to exert a number of important physiological activities such as antidiabetic⁹, antiemetic¹⁰, platelet aggregation inhibition, antihyperlipoproteinemic⁹, bronchodilatory, growth stimulatory¹¹, and antidepressant¹². They have also been used in the treatment of inflammatory diseases, pain, migraine, and asthma¹³.

Recently, we exploited the synthesis of 6-aryl-1,2,4,5-tetrazinane-3-thiones¹⁴, fused indazoles¹⁵, pyrimidinyl thiazolidin-4-ones¹⁶, and 2,6-diarylpiperidin-4-one derivatives^{17–19} with a view to incorporating various other

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bioactive heterocyclic nuclei such as 1,2,3-selenadiazoles, 1,2,3-thiadiazoles, and diazepans, intact, for the evaluation of associated antibacterial and antifungal activities. In view of the above and as part of the ongoing research on antimicrobials, we planned to synthesize a system that comprises both *N*-functionalized morpholine and 2-amino-4,6-dialkylpyrimidine components together to give a compact structure as the title 4-(4-morpholinophenyl)-6-arylpyrimidin-2-amines.

Experimental

Chemistry

Thin layer chromatography (TLC) was performed to assess the reactions and the purity of the products. All the reported melting points were taken in open capillaries and are uncorrected. Infrared (IR) spectra were recorded in KBr (pellet form) on a Nicolet Avatar-330 Fourier transform (FT)-IR spectrophotometer, and noteworthy absorption values (cm^{-1}) alone are listed. ^1H and ^{13}C nuclear magnetic resonance (NMR) spectra were recorded at 400 MHz and 100 MHz, respectively, on a Bruker AMX 400 NMR spectrometer using dimethylsulfoxide (DMSO)-*d* as solvent. Electrospray ionization (ESI)-+ve mass spectra (MS) were recorded on a Bruker Daltonics LC-MS spectrometer. Satisfactory microanalysis was obtained on a Carlo Erba 1106 CHN analyzer.

General procedure²⁰ for the synthesis of (E)-1-(4-morpholinophenyl)-3-aryl-prop-2-en-1-ones (11–19)

To an ethanolic solution of 1-(4-morpholinophenyl) ethanone (0.001 mol) and substituted benzaldehyde (0.001 mol), aqueous sodium hydroxide (0.005 mol) was added dropwise with stirring on a mechanical stirrer for 10 min, and stirring was continued for 4–6 h. After completion of reaction, the crude product isolated by suction was washed with water, dried, and recrystallized from ethanol.

(E)-1-(4-Morpholinophenyl)-3-phenyl-prop-2-en-1-one (11) Reaction time: 4 h; m.p.: 148–150°C; Yield: 95%; IR (KBr) ν (cm^{-1}): 3007, 2962, 2924, 2852, 1646, 1606, 1190, 769; MS: m/z 294 ($M + 1$)⁺; Molecular formula: $\text{C}_{19}\text{H}_{19}\text{NO}_2$; ^1H NMR (δ ppm): 3.33–3.36 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.87–3.89 (t, 4H, $\text{O}(\text{CH}_2)_2$), 6.93–6.95 (d, 1H, H_2 , $J=8.9$ Hz); 7.38–7.82 (m, 10H, H_{arom}); 8.01–8.03 (d, 1H, H_3 , $J=8.9$ Hz); ^{13}C NMR (δ ppm): 47.7 $\text{N}(\text{CH}_2)$, 66.5 $\text{O}(\text{CH}_2)$, 122.3 C-2, 143.2 C-3, 113.6, 128.8–130.3 $-\text{C}_{\text{arom}}$, 128.2, 135.4, 154.1 *ipso*-C, 188.1 C-1.

(E)-1-(4-Morpholinophenyl)-3-*p*-tolyl-prop-2-en-1-one (12) Reaction time: 5 h; m.p.: 178–180°C; Yield: 92%; IR (KBr) ν (cm^{-1}): 3012, 2923, 2924, 2851, 1645, 1600, 1194, 810; MS: m/z 308 ($M + 1$)⁺; Molecular formula: $\text{C}_{20}\text{H}_{21}\text{NO}_2$; ^1H NMR (δ ppm): 1.57 (s, 3H, CH_3 at phenyl ring), 3.32–3.35 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.86–3.89 (t, 4H, $\text{O}(\text{CH}_2)_2$), 6.92–6.94 (d, 1H, H_2 , $J=8.8$ Hz); 7.21–7.80 (m, 9H, H_{arom}); 8.00–8.02 (d, 1H, H_3 , $J=8.8$ Hz); ^{13}C NMR (δ ppm): 21.0 CH_3 at phenyl ring, 47.7 $\text{N}(\text{CH}_2)$, 66.5 $\text{O}(\text{CH}_2)$, 121.2 C-2, 143.3 C-3, 113.5, 129.3–130.5 $-\text{C}_{\text{arom}}$, 128.2, 132.7, 140.5, 154.1 *ipso*-C, 188.2 C-1.

(E)-3-(4-Chlorophenyl)-1-(4-morpholinophenyl)prop-2-en-1-one (13) Reaction time: 5 h; m.p.: 142–144°C; Yield: 90%;

IR (KBr) ν (cm^{-1}): 3087, 2967, 2920, 2859, 1597, 1654, 1202, 817; MS: m/z 328 ($M + 1$)⁺; Molecular formula: $\text{C}_{19}\text{H}_{18}\text{NO}_2\text{Cl}$; ^1H NMR (δ ppm): 3.34–3.37 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.89–3.91 (t, 4H, $\text{O}(\text{CH}_2)_2$), 6.97–6.99 (d, 1H, H_2 , $J=8.8$ Hz); 7.35–7.76 (m, 9H, H_{arom}); 8.00–8.02 (d, 1H, H_3 , $J=8.9$ Hz); ^{13}C NMR (δ ppm): 47.6 $\text{N}(\text{CH}_2)$, 66.5 $\text{O}(\text{CH}_2)$, 121.2 C-2, 141.8 C-3, 113.5, 129.4–130.6 $-\text{C}_{\text{arom}}$, 129.1, 133.8, 135.0, 154.0 *ipso*-C, 192.2 C-1.

(E)-3-(4-Methoxyphenyl)-1-(4-morpholinophenyl)prop-2-en-1-one (14) Reaction time: 4 h; m.p.: 110–112°C; Yield: 90%; IR (KBr) ν (cm^{-1}): 3010, 2961, 2918, 2841, 1645, 1601, 1225; MS: m/z 324 ($M + 1$)⁺; Molecular formula: $\text{C}_{20}\text{H}_{21}\text{NO}_3$; ^1H NMR (δ ppm): 3.32–3.35 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.87–3.90 (t, 4H, $\text{O}(\text{CH}_2)_2$), 3.86 (s, 3H, OCH_3 at phenyl ring), 7.59–7.61 (d, 1H, H_2 , $J=8.6$ Hz); 6.92–7.46 and 7.75–7.79 (m, 9H, H_{arom}); 8.00–8.02 (d, 1H, H_3 , $J=8.7$ Hz); ^{13}C NMR (δ ppm): 47.6 $\text{N}(\text{CH}_2)$, 55.3 OCH_3 at phenyl ring, 66.5 $\text{O}(\text{CH}_2)$, 119.6 C-2, 143.1 C-3, 113.5, 129.2–130.4 $-\text{C}_{\text{arom}}$, 127.9, 129.9, 153.9, 161.3 *ipso*-C, 187.8 C-1.

(E)-3-(4-Fluorophenyl)-1-(4-morpholinophenyl)prop-2-en-1-one (15) Reaction time: 4 h; m.p.: 159–161°C; Yield: 95%; IR (KBr) ν (cm^{-1}): 3009, 2969, 2919, 2849, 1650, 1602, 1227; MS: m/z 312 ($M + 1$)⁺; Molecular formula: $\text{C}_{19}\text{H}_{18}\text{NO}_2\text{F}$; ^1H NMR (δ ppm): 3.33–3.36 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.87–3.89 (t, 4H, $\text{O}(\text{CH}_2)_2$), 6.93–6.95 (d, 1H, H_2 , $J=8.9$ Hz); 7.08–7.78 (m, 9H, H_{arom}); 8.00–8.02 (d, 1H, H_3 , $J=8.9$ Hz); ^{13}C NMR (δ ppm): 47.5 $\text{N}(\text{CH}_2)$, 66.5 $\text{O}(\text{CH}_2)$, 121.6 C-2, 141.9 C-3, 113.4, 115.8, 130.1, 131.5 $-\text{C}_{\text{arom}}$, 128.8, 130.6, 154.1, 162.5 *ipso*-C, 187.8 C-1.

(E)-3-(4-Bromophenyl)-1-(4-morpholinophenyl)prop-2-en-1-one (16) Reaction time: 2 h; m.p.: 138–139°C; Yield: 94%; IR (KBr) ν (cm^{-1}): 3001, 2960, 2923, 2845, 1657, 1612, 1227; MS: m/z 372 ($M + 1$)⁺; Molecular formula: $\text{C}_{19}\text{H}_{18}\text{NO}_2\text{Br}$; ^1H NMR (δ ppm): 3.32–3.35 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.86–3.87 (t, 4H, $\text{O}(\text{CH}_2)_2$), 6.94–6.96 (d, 1H, H_2 , $J=8.8$ Hz); 7.18–7.82 (m, 9H, H_{arom}); 8.01–8.03 (d, 1H, H_3 , $J=8.6$ Hz); ^{13}C NMR (δ ppm): 47.9 $\text{N}(\text{CH}_2)$, 65.6 $\text{O}(\text{CH}_2)$, 121.8 C-2, 142.3 C-3, 113.8, 115.1, 130.7, 131.7 $-\text{C}_{\text{arom}}$, 128.5, 131.2, 154.7, 162.7 *ipso*-C, 188.8 C-1.

(E)-1-(4-Morpholinophenyl)-3-(3-nitrophenyl)prop-2-en-1-one (17) Reaction time: 6 h; m.p.: 134–316°C; Yield: 87%; IR (KBr) ν (cm^{-1}): 3087, 2966, 2923, 2862, 1651, 1608, 1224; MS: m/z 339 ($M + 1$)⁺; Molecular formula: $\text{C}_{19}\text{H}_{18}\text{N}_2\text{O}_4$; ^1H NMR (δ ppm): 3.36–3.38 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.88–3.90 (t, 4H, $\text{O}(\text{CH}_2)_2$), 6.95–6.97 (d, 1H, H_2 , $J=8.9$ Hz); 7.27–7.91 and 8.23–8.25 (m, 9H, H_{arom}); 8.03–8.05 (d, 1H, H_3 , $J=8.9$ Hz); ^{13}C NMR (δ ppm): 47.3 $\text{N}(\text{CH}_2)$, 66.9 $\text{O}(\text{CH}_2)$, 122.0 C-2, 140.1 C-3, 113.3, 124.2–134.2 $-\text{C}_{\text{arom}}$, 128.9, 137.1, 148.7, 154.3 *ipso*-C, 187.8 C-1.

(E)-1-(4-Morpholinophenyl)-3-(3-chlorophenyl)prop-2-en-1-one (18) Reaction time: 6 h; m.p.: 138–140°C; Yield: 90%; IR (KBr) ν (cm^{-1}): 3093, 2969, 2928, 2857, 1593, 1652, 1212, 830; MS: m/z 328 ($M + 1$)⁺; Molecular formula: $\text{C}_{19}\text{H}_{18}\text{NO}_2\text{Cl}$; ^1H NMR (δ ppm): 3.33–3.36 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.89–3.91 (t, 4H, $\text{O}(\text{CH}_2)_2$), 6.96–6.98 (d, 1H, H_2 , $J=8.9$ Hz); 7.33–7.81 (m, 9H, H_{arom}); 7.98–8.00 (d, 1H, H_3 , $J=8.7$ Hz); ^{13}C NMR (δ ppm): 47.8 $\text{N}(\text{CH}_2)$, 66.4 $\text{O}(\text{CH}_2)$, 121.2 C-2, 141.6 C-3, 113.3, 128.8–130.1 $-\text{C}_{\text{arom}}$, 129.3, 133.7, 145.2, 154.1 *ipso*-C, 192.3 C-1.

(E)-1-(4-Morpholinophenyl)-3-(3-fluorophenyl)prop-2-en-1-one (19) Reaction time: 5 h; m.p.: 152–157°C; Yield: 91%; IR (KBr) ν (cm^{-1}): 3018, 2974, 2924, 2843, 1649, 1605,

1226; MS: m/z 312 ($M + 1$)⁺; Molecular formula: C₁₉H₁₈NO₂F; ¹H NMR (δ ppm): 3.33–3.36 (t, 4H, N(CH₂)₂), 3.86–3.88 (t, 4H, O(CH₂)₂), 6.92–6.94 (d, 1H, H₂, $J = 8.8$ Hz); 7.18–7.68 (m, 9H, H_{arom}); 7.92–7.94 (d, 1H, H₃, $J = 8.7$ Hz); ¹³C NMR (δ ppm): 47.5 N(CH₂), 66.6 O(CH₂), 121.3 C-2, 141.8 C-3, 113.4, 125.3–130.1 -C_{arom}, 128.6, 130.5, 154.3, 162.7 *ipso*-C, 187.5 C-1.

General method for the synthesis of 4-(4-morpholinophenyl)-6-arylpyrimidin-2-amines (20–28)

A mixture of (*E*)-1-(4-morpholinophenyl)-3-aryl-prop-2-en-1-ones (**11–19**) (0.001 mol) and guanidine nitrate (0.001 mol) in ethanol (50 mL) was refluxed, while a solution of sodium hydroxide (0.005 mol) in water (10 mL) was added portion-wise for 2 h. Refluxing was continued for a further 10 h and the mixture was poured into ice-cold water. The formed solid was separated by filtration, and purified by column chromatography using silica gel (100–200 mesh), with ethyl acetate–petroleum ether (b.p. 40–60°C) in the ratio 2:8 as eluent.

4-(4-Morpholinophenyl)-6-phenylpyrimidin-2-amine (20) IR (KBr) (cm⁻¹): 3355, 3459, 3060, 2961, 2920, 1661, 1599, 1229, 928, 824, 776, 697, 634; ¹H NMR (δ ppm): 3.33–3.38 (t, 4H, N(CH₂)₂), 3.88–3.89 (t, 4H, O(CH₂)₂), 5.23 (s, 2H, NH₂), 7.38–7.85 (m, 10H, H_{arom}), the signal for H-5 proton may be merged with the aromatic protons; ¹³C NMR (δ ppm): 46.3 N(CH₂), 67.3 O(CH₂), 103.4 C-5, 163.8 C-2, 164.7 C-6, 165.0 C-4, 127.0–131.5 -C_{arom}, 142.1, 153.9 *ipso*-C.

4-(4-Morpholinophenyl)-6-*p*-tolylpyrimidin-2-amine (21) IR (KBr) (cm⁻¹): 3432, 3200, 2967, 2923, 1625, 1599, 1229, 928, 815, 645; ¹H NMR (δ ppm): 2.31 (s, 3H, CH₃), 3.34–3.38 (t, 4H, N(CH₂)₂), 3.86–3.89 (t, 4H, O(CH₂)₂), 5.25 (s, 2H, NH₂), 7.20–8.12 (m, 9H, H_{arom}), the signal for H-5 proton may be merged with the aromatic protons; ¹³C NMR (δ ppm): 25.4 CH₃, 46.7 N(CH₂), 67.5 O(CH₂), 104.1 C-5, 163.8 C-2, 164.1 C-6, 164.3 C-4, 126.0–131.4 -C_{arom}, 143.8, 152.4 *ipso*-C.

4-(4-Chlorophenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine (22) IR (KBr) (cm⁻¹): 3396, 3217, 3027, 2962, 2920, 1656, 1229, 930, 819, 684; ¹H NMR (δ ppm): 3.35–3.39 (t, 4H, N(CH₂)₂), 3.89–3.91 (t, 4H, O(CH₂)₂), 5.28 (s, 2H, NH₂), 7.26–7.85 (m, 9H, H_{arom}), the signal for H-5 proton may be merged with the aromatic protons; ¹³C NMR (δ ppm): 47.6 N(CH₂), 66.5 O(CH₂), 104.5 C-5, 163.8 C-2, 164.1 C-6, 164.7 C-4, 127.0–139.1 -C_{arom}, 141.5, 152.5 *ipso*-C.

4-(4-Methoxyphenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine (23) IR (KBr) (cm⁻¹): 3447, 3200, 2972, 2922, 1659, 1600, 1243, 929, 821, 607; ¹H NMR (δ ppm): 3.32–3.35 (t, 4H, N(CH₂)₂), 3.86 (s, 3H, OCH₃), 3.87–3.90 (t, 4H, O(CH₂)₂), 5.26 (s, 2H, NH₂), 7.18–7.82 (m, 9H, H_{arom}), the signal for H-5 proton may be merged with the aromatic protons; ¹³C NMR (δ ppm): 46.7 N(CH₂), 55.0 OCH₃, 67.3 O(CH₂), 103.9 C-5, 163.7 C-2, 164.7 C-6, 165.0 C-4, 127.5–142.1 -C_{arom}, 153.8, 154.2 *ipso*-C.

4-(4-Fluorophenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine (24) IR (KBr) (cm⁻¹): 3434, 3200, 2967, 2923, 1624, 1599, 1226, 928, 815, 645; ¹H NMR (δ ppm): 3.34–3.36 (t, 4H, N(CH₂)₂), 3.87–3.88 (t, 4H, O(CH₂)₂), 5.29 (s, 2H, NH₂), 7.30–8.03 (m, 9H, H_{arom}), the signal for H-5 proton may be

merged with the aromatic protons; ¹³C NMR (δ ppm): 46.6 N(CH₂), 67.3 O(CH₂), 103.5 C-5, 163.8 C-2, 164.2 C-6, 164.4 C-4, 126.5–140.0 -C_{arom}, 140.6, 154.0 *ipso*-C.

4-(4-Bromophenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine (25) IR (KBr) (cm⁻¹): 3398, 3219, 3029, 2965, 2922, 1659, 1231, 934, 821, 687; ¹H NMR (δ ppm): 3.36–3.39 (t, 4H, N(CH₂)₂), 3.87–3.91 (t, 4H, O(CH₂)₂), 5.29 (s, 2H, NH₂), 7.29–7.87 (m, 9H, H_{arom}), the signal for H-5 proton may be merged with the aromatic protons; ¹³C NMR (δ ppm): 47.8 N(CH₂), 66.6 O(CH₂), 104.8 C-5, 163.9 C-2, 164.3 C-6, 164.8 C-4, 127.5–139.7 -C_{arom}, 152.6, 141.6 *ipso*-C.

4-(4-Morpholinophenyl)-6-(3-nitrophenyl)pyrimidin-2-amine (26) IR (KBr) (cm⁻¹): 3400, 3200, 3060, 2961, 2920, 1661, 1566, 1229, 928, 824, 776, 697; ¹H NMR (δ ppm): 3.36–3.39 (t, 4H, N(CH₂)₂), 3.88–3.92 (t, 4H, O(CH₂)₂), 5.28 (s, 2H, NH₂), 7.27–8.28 (m, 9H, H_{arom}), the signal for H-5 proton may be merged with the aromatic protons; ¹³C NMR (δ ppm): 47.3 N(CH₂), 67.6 O(CH₂), 104.4 C-5, 163.8 C-2, 164.2 C-6, 164.4 C-4, 125.2–131.5 -C_{arom}, 146.7, 153.9 *ipso*-C.

4-(3-Chlorophenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine (27) IR (KBr) (cm⁻¹): 3398, 3219, 3028, 2963, 2923, 1655, 1228, 929, 817, 686; ¹H NMR (δ ppm): 3.36–3.38 (t, 4H, N(CH₂)₂), 3.88–3.90 (t, 4H, O(CH₂)₂), 5.29 (s, 2H, NH₂), 7.16–7.74 (m, 9H, H_{arom}), the signal for H-5 proton may be merged with the aromatic protons; ¹³C NMR (δ ppm): 47.7 N(CH₂), 66.6 O(CH₂), 104.4 C-5, 163.6 C-2, 164.3 C-6, 164.8 C-4, 126.2–138.8 -C_{arom}, 146.5, 152.3 *ipso*-C.

4-(3-Fluorophenyl)-6-(4-morpholinophenyl)pyrimidin-2-amine (28) IR (KBr) (cm⁻¹): 3437, 3204, 2965, 2928, 1623, 1597, 1225, 921, 811, 649; ¹H NMR (δ ppm): 3.33–3.35 (t, 4H, N(CH₂)₂), 3.87–3.88 (t, 4H, O(CH₂)₂), 5.27 (s, 2H, NH₂), 7.28–8.05 (m, 9H, H_{arom}), the signal for H-5 proton may be merged with the aromatic protons; ¹³C NMR (δ ppm): 46.8 N(CH₂), 67.4 O(CH₂), 103.7 C-5, 163.9 C-2, 164.4 C-6, 164.5 C-4, 125.9–140.3 -C_{arom}, 146.6, 154.2 *ipso*-C.

Microbiology

Materials

All the clinically isolated bacterial strains, namely *Staphylococcus aureus*, β -hemolytic *Streptococcus*, *Vibrio cholerae*, *Shigella flexneri*, *Klebsiella pneumoniae*, and *Pseudomonas aeruginosa* and fungal strains, namely *Aspergillus flavus*, *Candida albicans*, *Mucor*, and *Rhizopus* were obtained from the Faculty of Medicine, Annamalai University, Annamalai Nagar, Tamil Nadu, India.

In vitro antibacterial and antifungal activity

Minimum inhibitory concentration (MIC) values in $\mu\text{g/mL}$ were determined by the two-fold serial dilution method²¹. The respective test compounds (**20–28**) were dissolved in DMSO to obtain 1 mg mL⁻¹ stock solution. Seeded broth (broth containing microbial spores) was prepared in nutrient broth (NB) from 24-h-old bacterial cultures on nutrient agar (Hi-media, Mumbai) at 37 \pm 1°C, while fungal spores from 1- to 7-day-old Sabouraud agar (Hi-media) slant cultures were suspended in Sabouraud dextrose broth (SDB). The colony forming units (cfu) of the seeded broth were determined by

the plating technique and adjusted in the range of 10^4 – 10^5 cfu/mL. The final inoculum size was 10^5 cfu/mL for the antibacterial assay and 1.1 – 1.5×10^2 cfu/mL for the antifungal assay. Testing was performed at $\text{pH } 7.4 \pm 0.2$ for bacteria (NB) and at $\text{pH } 5.6$ for fungi (SDB). Exactly 0.4 mL of test compound solution was added to 1.6 mL of seeded broth to form the first dilution. One milliliter of this was diluted with a further 1 mL of seeded broth to give the second dilution, and so on till six such dilutions were obtained. A set of assay tubes containing only seeded broth were kept as control. The tubes were incubated in biochemical oxygen demand (BOD) incubators at $37 \pm 1^\circ\text{C}$ for bacteria and $28 \pm 1^\circ\text{C}$ for fungi. The minimum inhibitory concentrations (MICs) were recorded by visual observation after 24 h (for bacteria) and 72–96 h (for fungi) of incubation. Ciprofloxacin was used as the standard for bacterial studies and fluconazole was used as the standard for fungal studies.

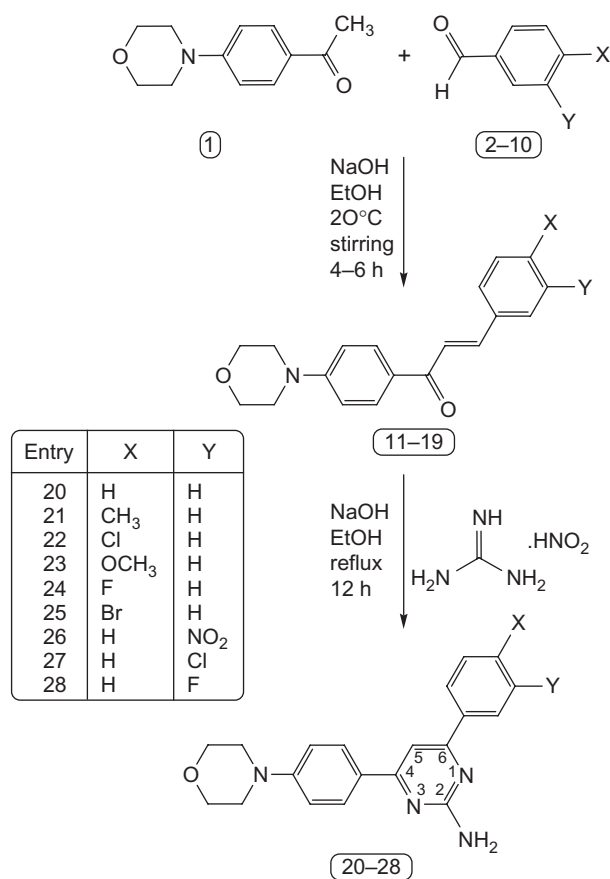
Results and discussion

Chemistry

(*E*)-1-(4-Morpholinophenyl)-3-aryl-prop-2-en-1-ones (**11–19**) were synthesized by stirring commercially available 1-(4-morpholinophenyl) ethanone and substituted benzaldehyde in the presence of sodium hydroxide in ethanol at 20°C for 1 h. Treatment of compounds **11–19** with guanidine nitrate²² in the presence of sodium hydroxide in refluxing ethanol for 8 h yielded the respective 4-(4-morpholinophenyl)-6-aryl-pyrimidin-2-amines (**20–28**). A schematic representation and analytical data of compounds **20–28** are shown in Scheme 1 and Table 1, respectively. The structure of the newly synthesized compounds **20–28** was confirmed by the melting points, elemental analysis, and MS, FT-IR, and one-dimensional NMR (^1H and ^{13}C) spectroscopic data.

Antibacterial activity

Novel 4-(4-morpholinophenyl)-6-aryl-pyrimidin-2-amines (**20–28**) were tested for their antibacterial activity *in vitro* against *S. aureus*, β -hemolytic *Streptococcus*, *V. cholerae*, *S. felxneri*, *K. pneumoniae*, and *P. aeruginosa*. Ciprofloxacin was used as the standard drug. Minimum inhibitory concentration (MIC) values are reproduced in Table 2. Compound **20**, which had no substitution at the *para* position of the phenyl ring, only exerted moderate activities against all the used bacterial strains except against *V. cholerae*, with pronounced activity at $12.5 \mu\text{g/mL}$, besides compounds **23**, **25**, **27**, and **28**, which were also potently active. Compounds **21** (against *S. felxneri*) and **23** (against β -hemolytic *Streptococcus*) did not exhibit antibacterial activity even at higher concentrations, i.e. $200 \mu\text{g/mL}$. Compounds **22**, **25**, **26**, and **28**, which contained electron-withdrawing chloro, bromo, nitro, and fluoro functional groups, respectively, at the *para/meta* position of the phenyl ring attached to the pyrimidine ring promoted much activity against *S. aureus*. Compounds **24** and **25** (against β -hemolytic *Streptococcus*) and compound **28** (against *S. felxneri*) showed pronounced activity.



Scheme 1. Synthesis reaction pathway for formation of 4-(4-morpholinophenyl)-6-aryl-pyrimidin-2-amines.

Compounds **26** and **28** (against *K. pneumoniae*) and compounds **24**, **25**, and **28** (against *P. aeruginosa*) exerted strong antibacterial activity.

Antifungal activity

The *in vitro* antifungal activity of 4-(4-morpholinophenyl)-6-aryl-pyrimidin-2-amines (**20–28**) was studied against the fungal strains *A. flavus*, *C. albicans*, *Mucor*, and *Rhizopus*. Fluconazole was used as the standard drug. The MIC values are reproduced in Table 3. Compound **20**, which had no substitution at the *para* position of the phenyl ring, compound **21**, which had an electron-donating CH₃ group, and compound **26**, which had an electron-withdrawing NO₂ group attached at the *meta* position of the phenyl ring attached to the pyrimidine ring, were less active against all tested fungal strains, whereas compound **27** did not exhibit antifungal activity against *A. flavus* and *C. albicans*. Further introduction of an electron-withdrawing chloro/bromo functional group at the *para* position of the phenyl ring attached to the pyrimidine ring in compounds **22** and **25** promoted much activity against *A. flavus*, while against *Mucor* compounds **24** and **25** registered maximum activity. Compound **23**, which had an electron-donating methoxy group at the *para* position of the phenyl ring attached to the pyrimidine ring, was more potent against *C. albicans*. Instead, where the chloro/fluoro group was attached at the *meta* position of the phenyl

Table 1. Physical and analytical data of compounds **11–28**.

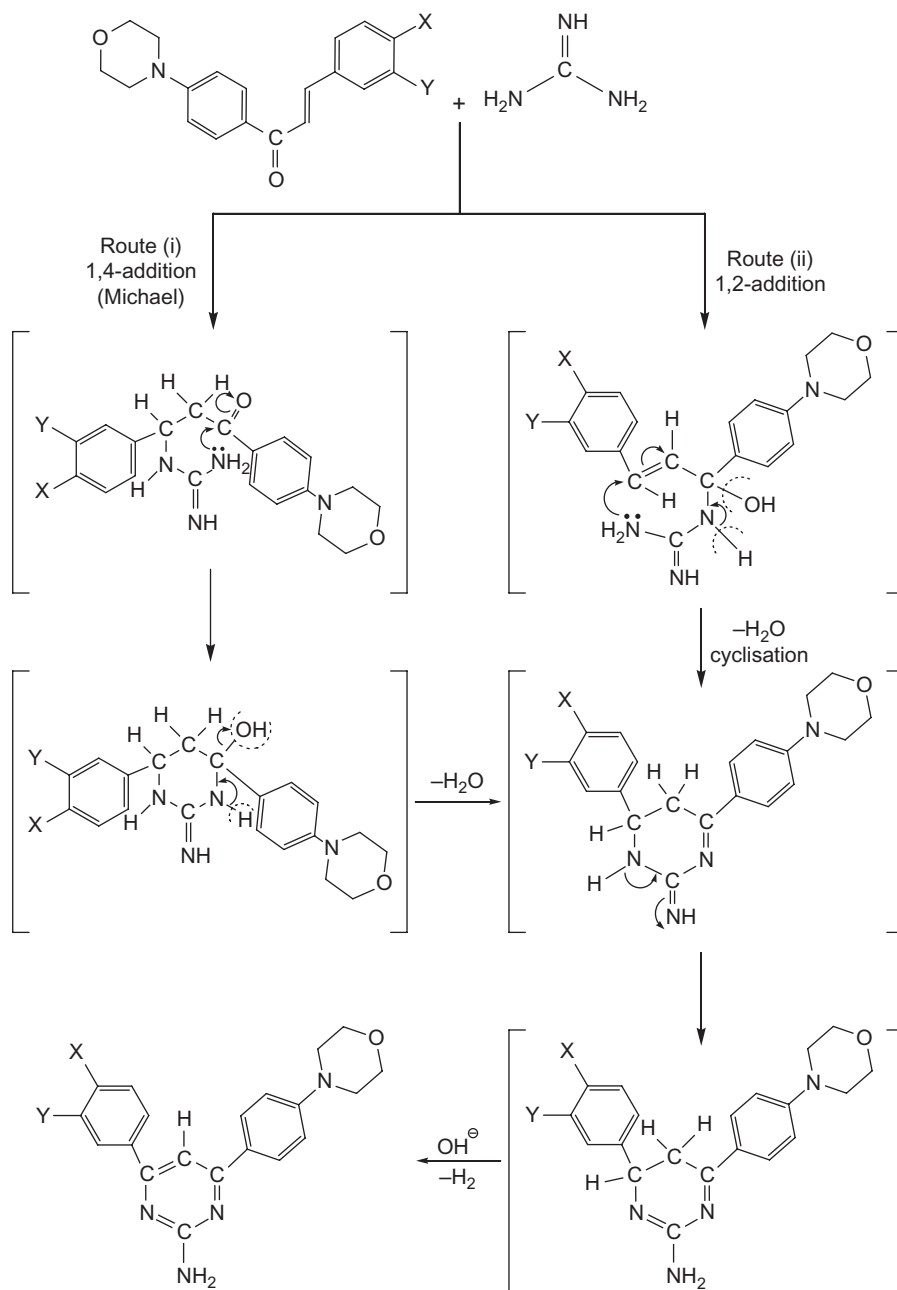
Entry	X	Y	Yield (%)	m.p. (°C)	Elemental analysis (%)			<i>m/z</i> (M + 1) ⁺ , molecular formula
					C, Calc. (Found)	H, Calc. (Found)	N, Calc. (Found)	
11	H	H	95	149	77.81 (77.78)	6.48 (6.46)	4.77 (4.76)	294 C ₁₉ H ₁₉ NO ₂
12	CH ₃	H	92	179	78.17 (78.15)	6.84 (6.81)	4.56 (4.54)	308 C ₂₀ H ₂₁ NO ₂
13	Cl	H	90	143	69.72 (69.70)	5.50 (5.47)	4.28 (4.26)	328 C ₁₉ H ₁₆ NO ₂ Cl
14	OCH ₃	H	90	111	74.30 (74.29)	6.50 (6.48)	4.33 (4.31)	324 C ₂₀ H ₂₁ NO ₃
15	F	H	95	160	73.31 (73.29)	5.78 (5.76)	4.50 (4.48)	312 C ₁₉ H ₁₈ NO ₂ F
16	Br	H	85	145	61.30 (61.27)	4.87 (4.83)	3.76 (3.74)	372 C ₁₉ H ₁₈ BrNO ₂
17	H	NO ₂	87	135	67.45 (67.43)	5.32 (5.29)	8.28 (8.26)	339 C ₁₉ H ₁₈ N ₂ O ₄
18	H	Cl	90	138	69.72 (69.69)	5.50 (5.48)	4.28 (4.25)	328 C ₁₉ H ₁₆ NO ₂ Cl
19	H	F	90	154	73.31 (73.28)	5.78 (5.74)	4.50 (4.47)	312 C ₁₉ H ₁₈ NO ₂ F
20	H	H	80	118	72.09 (72.05)	6.30 (6.28)	16.80 (16.77)	333 C ₂₀ H ₂₁ N ₄ O
21	CH ₃	H	85	73	72.64 (72.61)	6.62 (6.59)	16.12 (16.10)	347 C ₂₁ H ₂₃ N ₄ O
22	Cl	H	78	123	65.33 (65.31)	5.44 (5.41)	15.23 (15.20)	367 C ₂₀ H ₂₀ N ₄ OCl
23	OCH ₃	H	80	91	69.44 (69.41)	6.33 (6.31)	15.41 (15.38)	363 C ₂₁ H ₂₃ N ₄ O ₂
24	F	H	90	87	68.40 (68.38)	5.69 (5.66)	15.94 (15.91)	351 C ₂₀ H ₂₀ N ₄ OF
25	Br	H	90	93	58.40 (58.36)	4.66 (4.64)	13.62 (13.59)	411 C ₂₀ H ₁₉ BrN ₄ O
26	H	NO ₂	85	176	65.96 (65.92)	5.49 (5.47)	15.38 (15.35)	364 C ₂₀ H ₂₀ N ₅ O ₃
27	H	Cl	80	135	65.33 (65.30)	5.44 (5.42)	15.23 (15.20)	367 C ₂₀ H ₂₀ N ₄ OCl
28	H	F	78	96	68.40 (68.37)	5.69 (5.68)	15.94 (15.92)	351 C ₂₀ H ₂₀ N ₄ OF

ring attached to the pyrimidine ring, **27** and **28** promoted good activity against *Rhizopus*.

Conclusion

Microbiological screening studies carried out to evaluate the antibacterial and antifungal potencies of the newly synthesized 4-(4-morpholinophenyl)-6-aryl-pyrimidin-2-amines (**20–28**) are clearly illustrated by Tables 2 and 3. Close inspection of the *in vitro* antibacterial and antifungal activity profiles in different electron-donating (CH₃ and OCH₃) and electron-withdrawing (-F, -Cl, -Br, and -NO₂) functional group-substituted phenyl rings of the novel 4-(4-morpholinophenyl)-6-aryl-pyrimidin-2-amines (**20–28**) showed that they exerted strong antibacterial activity against all tested bacterial strains. Compounds **20**, **23**, **25**, **27**, and **28** exerted excellent antibacterial activity against

V. cholerae, and compounds **21** (against *S. felxneri*) and **23** (against β-hemolytic *Streptococcus*) did not exhibit antibacterial activity even at higher concentrations, i.e. 200 μg/mL. Compounds **22**, **25**, **26**, and **28**, which contained electron-withdrawing chloro, bromo, nitro, and fluoro functional groups, respectively, at the *para/meta* position of the phenyl ring attached to the pyrimidine ring promoted much activity against *S. aureus*. Compounds **24** and **25** (against β-hemolytic *Streptococcus*) and compound **28** (against *S. felxneri*) showed pronounced activity. Compounds **26** and **28** (against *K. pneumoniae*) and compounds **24**, **25**, and **28** (against *P. aeruginosa*) exerted strong antibacterial activity. Results of the antifungal activity study showed that the nature of the substituent on the phenyl ring, namely chloro, fluoro, and bromo, and methoxy functions at the *para/meta* positions of the aryl moieties, were determinants of the nature and extent of the antifungal activity of



Scheme 2. Plausible reaction mechanism for formation of target molecules.

Table 2. *In vitro* antibacterial activity (MIC) values for compounds 20–28.

Compound	X	Y	Minimum inhibitory concentration (MIC) ($\mu\text{g}/\text{mL}$)					
			<i>S. aureus</i>	β -Hemolytic <i>Streptococcus</i>	<i>V. cholerae</i>	<i>S. felxneri</i>	<i>K. pneumoniae</i>	<i>P. aeruginosa</i>
20	H	H	100	100	12.5	100	200	25
21	CH_3	H	50	100	100	— ^a	100	100
22	Cl	H	6.25	50	100	100	50	50
23	OCH_3	H	100	— ^a	6.25	100	50	50
24	F	H	100	12.5	200	100	100	6.25
25	Br	H	6.25	25	12.5	25	25	12.5
26	H	NO_2	12.5	50	200	100	6.25	50
27	H	Cl	25	100	50	50	100	12.5
28	H	F	12.5	25	12.5	6.25	6.25	25
Ciprofloxacin	—	—	25	50	25	50	50	25

^aNo inhibition even at higher concentration, i.e. 200 $\mu\text{g}/\text{mL}$.

Table 3. *In vitro* antifungal activity (MIC) values for compounds 20–28.

Compound	X	Y	Minimum inhibitory concentration (MIC) ($\mu\text{g/mL}$)			
			<i>A. flavus</i>	<i>C. albicans</i>	<i>Mucor</i>	<i>Rhizopus</i>
20	H	H	50	100	25	200
21	CH ₃	H	50	200	100	25
22	Cl	H	6.25	100	200	100
23	OCH ₃	H	100	6.25	25	50
24	F	H	200	100	6.25	100
25	Br	H	6.25	25	12.5	12.5
26	H	NO ₂	100	100	100	100
27	H	Cl	— ^a	— ^a	12.5	6.25
28	H	F	50	100	25	12.5
Fluconazole	—	—	25	50	50	50

^aNo inhibition even at higher concentration, i.e. 200 $\mu\text{g/mL}$.

all synthesized compounds 20–28 on the fungal strains *A. flavus*, *C. albicans*, *Mucor*, and *Rhizopus*. The introduction of an electron-withdrawing chloro/bromo functional group at the *para* position of the phenyl ring attached to the pyrimidine ring in compounds 22 and 25 promoted much activity against *A. flavus*, while compounds 24 and 25 registered maximum activity against *Mucor*. Compound 23, which had an electron-donating methoxy group at the *para* position of the phenyl ring attached to the pyrimidine ring, was more potent against *C. albicans*. Instead, when the chloro/fluoro group was attached at the *meta* position of the phenyl ring attached to the pyrimidine ring, compounds 27 and 28 promoted pronounced activity against *Rhizopus*. The method of action of these compounds is unknown. These observations may promote progress of our research in this field. Further development of this group of 4-(4-morpholinophenyl)-6-aryl-pyrimidin-2-amines (20–28) may lead to compounds with better pharmacological profiles than those of standard antibacterial and antifungal drugs.

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